

Growth-dependent changes in elemental stoichiometry and macromolecular allocation in the coccolithophore *Emiliania huxleyi* under different environmental conditions

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Abstract

The growth rate hypothesis (GRH) posits an increase in ribosomal ribonucleic acid (RNA) content, and therefore cellular phosphorus (P), with increasing growth rate. There is evidence that the GRH may not apply to phytoplankton under all conditions. Here, we experimentally controlled four conditions (light, temperature, pH, and CO₂) to alter the growth rate of *Emiliania huxleyi*, a biogeochemically important coccolithophorid, and monitored changes in RNA, protein, and carbohydrate content. We show that an increase in growth rate caused by increasing light, pH, and CO₂ resulted in increased RNA per unit of organic carbon (RNA : POC), but that increasing temperature, leading to increase of growth rate, resulted in a decrease in RNA : POC. Protein per unit of organic carbon (protein : POC) increased in our increased temperature, pH, and CO₂ treatments that increased growth rate, but there was little change in protein : POC in our light treatment despite it inducing the same increase in growth rate. Carbohydrate per unit of organic carbon (Carbohydrate : POC) increased with growth rate under increased light and CO₂ but did not vary significantly in the temperature or pH treatments. These results indicate that physiological acclimation to specific environmental conditions can lead to contrasting patterns in RNA, protein, and carbohydrate composition and therefore contrasting changes in carbon : nitrogen : phosphorus ratios with growth rate in *E. huxleyi*.

Alfred Redfield found that the molar ratio of carbon (C), nitrogen (N), and phosphorus (P) of marine plankton is often approximately 106 : 16 : 1 (Redfield 1958). There is increasing recognition that the C : N : P of plankton varies widely across regions and seasons and is associated with differences in community composition, nutrient supply rates, and environmental conditions (Finkel et al. 2010, 2016; Martiny et al. 2013, 2016). The C : N : P of plankton affects the C, N, and P cycles in the ocean (Redfield 1958; Weber and Deutsch 2010). Thus, quantifying the impact of marine environmental drivers on

the elemental stoichiometry of key plankton groups will improve our ability to model the biogeochemical cycles of C, N, and P (Arrigo 2005; Daines et al. 2014; Moreno and Martiny 2018).

The growth rate hypothesis (GRH) posits that growth rate will regulate elemental composition through shifts in cellular protein and RNA content. Specifically, Elser et al. (2000) propose that higher growth rates require a larger investment in P-rich ribosomes (RNA) to increase protein synthesis rates. As a consequence, increases in growth rate are expected to be associated with higher RNA : POC, RNA : protein, and therefore lower N : P content (Elser et al. 2000; Finkel et al. 2010; Moreno and Martiny 2018). The GRH and resulting impact on C : N : P assumes that ribosome activity per unit of ribosomal RNA (rRNA) is relatively constant and P allocated to rRNA constitutes the majority of the cellular P content (Elser et al. 2003; Flynn et al. 2010). The applicability of the GRH to phytoplankton has been questioned (Flynn et al. 2010; Garcia et al. 2016; Moreno and Martiny 2018). One of the reasons that growth rate may not be correlated with N : P is that

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phytoplankton are capable of luxury P consumption and can store large amounts of P (Dyhrman et al. 2006; Orchard et al. 2010; Diaz et al. 2016). In addition, individual environmental drivers may uniquely alter macromolecular allocation (protein, carbohydrate, RNA, pigments) and C : N : P in phytoplankton (Harris et al. 2009; Giordano et al. 2015; Liefer et al. 2019). Much of the work that supports the GRH was conducted on freshwater organisms and growth was often controlled by phosphorus, which will have a direct impact on cellular P content and P-rich cellular machinery (Elser et al. 2003; Nicklisch and Steinberg 2009; Flynn et al. 2010). There are few studies comparing the applicability of the GRH in phytoplankton under a range of environmental conditions (Giordano et al. 2015; Garcia et al. 2016) such as irradiance, temperature, carbon dioxide (CO₂) concentration, and pH.

Coccolithophores are biogeochemically important constituents of phytoplankton communities that produce both particulate organic carbon (POC) via photosynthesis and particulate inorganic carbon (PIC) through calcification (Rost and Riebesell 2004). Within subtropical gyres coccolithophores may contribute 20% or more to total organic carbon fixation (Poulton et al. 2007), and about 50% of CaCO₃ export to tropical and higher-latitude sediments (Broecker and Clark 2009), and hence they play an important role in the marine biological C pump and the biogeochemical cycling of carbon. The cosmopolitan coccolithophore *Emiliania huxleyi* is able to form massive blooms in temperate and subpolar waters (Iida et al. 2012; Kondrik et al. 2019). Therefore, it is of interest to improve our biochemical understanding of how the biogeographically dominant *E. huxleyi*, alter both carbon fixation and C : N : P under a range of environmental conditions (Riebesell and Tortell 2011; Rivero-Calle et al. 2015; Krumhardt et al. 2017).

Carbon dioxide concentration and pH appear to have their own unique physiological effects on coccolithophores; CO₂ concentration mainly influences photosynthetic carbon fixation and carbohydrate synthesis, while pH alters cellular pH homeostasis and ion balance, with downstream impacts on enzyme activity and C, N, and P metabolism (Hinga 2002; Taylor et al. 2011; Bach et al. 2013). There is not enough data available to fully evaluate how CO₂ or pH will alter RNA and protein content and shift C : N : P in coccolithophores (Engel et al. 2014; Olson et al. 2017) (Table S1). Increasing irradiance, below supersaturating levels, generally causes an increase in carbon storage and downregulation of light harvesting pigment-protein complexes in phytoplankton, often leading to higher C : N and lower N : P with increasing growth rates (Leonardos and Geider 2005; Jin et al. 2017; Feng et al. 2018) (Table S1). In contrast, in response to lowered temperature, organisms sometimes increase cellular RNA content, resulting in lower C : P and N : P (Feng et al. 2018; Wang et al. 2019). To improve our understanding of how environmental conditions influence macromolecular and elemental composition and to test the GRH in the coccolithophore *E. huxleyi* over a

range of conditions we quantify how C : N : P, RNA, and protein content vary with growth rate at 28, 45, and 280 $\mu\text{mol photons m}^{-2} \text{ s}^{-1}$; at 11.0°C, 14.5°C, and 22.0°C; at pH 6.70, 7.40, and 8.04; and at 2.70, 6.40, and 41.10 Pa.

Methods

Experimental setup

E. huxleyi strain RCC1266 (morphotype A), originally isolated from shelf waters around Ireland (49°30'N, 10°30' W), was obtained from the Roscoff algal culture collection. In the control treatment, *E. huxleyi* cells were maintained in 280 $\mu\text{mol photons m}^{-2} \text{ s}^{-1}$ under a 12 h : 12 h light : dark cycle (light period: 07:00 to 19:00 h), 22.0°C, pH_{Total} (total scale) 8.04, and 41.10 Pa (1 Pa \approx 10 μatm) pCO₂ in semicontinuous cultures. We defined the maximum growth rate (μ_{max}) as the steady-state exponential growth rate under the control conditions in this experiment. Experimental treatments include two subsaturating irradiances (28 and 45 $\mu\text{mol photons m}^{-2} \text{ s}^{-1}$), two lower suboptimal temperatures (11.0°C and 14.5°C), two lower pH (6.70 and 7.40), and two lower CO₂ concentrations (2.70 and 6.40 Pa), chosen to reduce growth rate to approximately $\frac{3}{4}\mu_{\text{max}}$ and $\frac{1}{2}\mu_{\text{max}}$ relative to the control (see Table 1 for a summary). Please note that the experimental conditions in the control treatment and the maximum growth rate are the same for all treatments. During pre-experiments, we investigated the growth response of *E. huxleyi* to a broad range of light intensities, temperatures, pH values, and CO₂ concentrations. Changing one condition at a time, we found that growth rates were close to $\frac{3}{4}\mu_{\text{max}}$ at 45 $\mu\text{mol photons m}^{-2} \text{ s}^{-1}$, 14.5°C, pH 7.40, and 6.40 Pa, and close to $\frac{1}{2}\mu_{\text{max}}$ at 28 $\mu\text{mol photons m}^{-2} \text{ s}^{-1}$, 11.0°C, pH 6.70, and 2.70 Pa. The artificial seawater (ASW) media was prepared according to Berges et al. (2001) without the addition of NaHCO₃, with a salinity of 33 psu. The dissolved inorganic carbon (DIC)-free ASW media was enriched with 100 $\mu\text{mol L}^{-1} \text{ NO}_3^-$, 6.25 $\mu\text{mol L}^{-1} \text{ PO}_4^{3-}$, 4.24 $\mu\text{mol L}^{-1} \text{ SiO}_3^{2-}$, 1/4 concentrations for trace elements, and f/8 concentrations for vitamins (Guillard and Ryther 1962; Guillard and Hargraves 1993). The carbonate chemistry of the enriched media was adjusted by adding 1.18 mol kg⁻¹ Na₂CO₃ and 2.00 mol kg⁻¹ hydrochloric acid (HCl) to achieve target pH_{Total} (total scale) and CO₂ levels (see below for more detail).

In the light and temperature treatments (four replicates), carbonate chemistry was preadjusted by stepwise additions of Na₂CO₃ and HCl. For the pH 7.40 and 6.70 treatments, calculated concentrations of Na₂CO₃ were added into the enriched ASW to maintain similar pCO₂ levels (about 41.00 Pa). Seawater pH was then adjusted to 7.40 or 6.70 by adding 5 mmol L⁻¹ of 2-[4-(2-hydroxyethyl)-1-piperazinyl]-ethanesulfonic acid (HEPES). For the 6.40 and 2.70 Pa CO₂ treatments, calculated amounts of Na₂CO₃ were added to the enriched ASW (to reach the pre-described growth rates), and pH was adjusted to 8.04 by adding 5 mmol L⁻¹ HEPES. There were four replicates at pH 7.40 and

Table 1. Measured light intensity, temperature, carbonate chemistry parameters at the start and end of the incubation, and changes in carbonate chemistry parameters during the incubation. Data are one replicate at the start of incubation and means \pm SD of four replicates at the end of the incubation. L280, L45, L28, T14.5, T11.0, pH 7.40, pH 6.70, C6.40, and C2.70 indicate that samples are, respectively, cultured under 280, 45, or 28 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$, under 14.5°C or 11.0°C, under pH 7.40 or 6.70, under 6.40 or 2.70 Pa CO_2 conditions.

Treatment		Light ($\mu\text{mol photons m}^{-2} \text{s}^{-1}$)	Temperature (°C)	$p\text{CO}_2$ (Pa)	pH (total scale)	TA ($\mu\text{mol kg}^{-1}$)	DIC ($\mu\text{mol kg}^{-1}$)	HCO_3^- ($\mu\text{mol kg}^{-1}$)	CO_3^{2-} ($\mu\text{mol kg}^{-1}$)	CO_2 ($\mu\text{mol kg}^{-1}$)
Control	L280	Start	22.0	41.05	8.04	2278	2002	1791	198.70	12.70
		End		32.62 \pm 0.62	8.11 \pm 0.01	2191 \pm 1	1885 \pm 4	1660 \pm 6	214.91 \pm 2.58	10.09 \pm 0.19
Light		Change		20.53%	6.69%	3.84%	5.86%	7.31%	8.13%	20.53%
	L45	Start	22.0	41.05	8.04	2278	2002	1791	198.70	12.70
		End		34.59 \pm 0.5	8.10 \pm 0.01	2226 \pm 4	1925 \pm 4	1701 \pm 5	212.95 \pm 1.93	10.70 \pm 0.20
		Change		15.73%	5.22%	2.30%	3.85%	4.99%	7.14%	15.73%
	L28	Start	22.0	41.05	8.04	2278	2002	1791	198.70	12.70
		End		31.99 \pm 0.7	8.11 \pm 0.01	2198 \pm 2	1886 \pm 5	1658 \pm 6	218.65 \pm 2.91	9.89 \pm 0.20
Temperature		Change		22.06%	7.49%	3.53%	5.77%	7.41%	10.01%	22.06%
	T14.5	Start	14.5	42.42	8.03	2277	2067	1899	152.10	16.30
		End		32.45 \pm 1.26	8.10 \pm 0.01	2111 \pm 1	1879 \pm 7	1706 \pm 10	160.57 \pm 4.24	12.43 \pm 0.48
		Change		23.51%	7.00%	7.27%	9.11%	10.16%	5.57%	23.51%
	T11.0	Start	11.0	41.03	8.04	2279	2090	1936	137.00	17.60
		End		35.53 \pm 1.01	8.10 \pm 0.01	2293 \pm 4	2078 \pm 6	1909 \pm 8	153.93 \pm 3.07	15.20 \pm 0.43
pH		Change		13.40%	5.66%	6.30%	0.57%	1.37%	12.38%	13.40%
	pH 7.40	Start	22.0	40.80	7.40		431	408	10.40	12.60
		End		32.30 \pm 1.20	7.40 \pm 0.01		344 \pm 10	326 \pm 10	8.40 \pm 0.30	9.97 \pm 0.40
		Change		20.91%	0.50%		20.01%	20.00%	19.06%	20.91%
	pH 6.70	Start	22.0	41.30	6.70		96	82	0.40	12.80
		End		24.50 \pm 0.60	6.72 \pm 0.01		58 \pm 2	51 \pm 2	0.27 \pm 0.01	7.60 \pm 0.18
CO ₂		Change		40.73%	1.50%		38.91%	38.63%	36.47%	40.73%
	C6.40	Start	22.0	6.40	8.04		312	279	31.00	2.00
		End		4.60 \pm 0.10	8.08 \pm 0.01		248 \pm 4	220 \pm 4	26.90 \pm 0.50	1.40 \pm 0.04
		Change		28.55%	4.25%		20.45%	21.21%	13.11%	28.55%
	C2.70	Start	22.0	2.70	8.04		130	116	12.90	0.80
		End		1.80 \pm 0.10	8.08 \pm 0.01		98 \pm 2	87 \pm 1	10.60 \pm 0.30	0.57 \pm 0.00
		Change		31.07%	3.75%		24.21%	24.85%	18.05%	31.07%

6.40 Pa CO₂ treatments, and eight replicates at pH 6.70 and 2.70 Pa CO₂ treatments. Under the light, temperature, and pH treatments, the seawater was placed in the incubator at the appropriate experimental temperature and bubbled for 24 h with filter-sterilized air pumped from the room. The dry air was humidified with Milli-Q water prior to the aeration to minimize evaporation. Under the CO₂ treatments, the ASW was not bubbled but was put into the incubator at 22.0°C for 24 h prior to inoculation.

After measuring pH_{Total} the media was filtered (0.22 µm pore size, Polycap 75 AS, Whatman) and carefully pumped into autoclaved 250 mL (for total alkalinity [TA] measurements under the light and temperature treatments, for DIC measurements under the pH and CO₂ treatments), 2300 mL (for pre-experimental cultures), and 4630 mL (for experimental cultures) polycarbonate (PC) bottles with no headspace to minimize gas exchange. The volumes of culture inoculum were calculated to match the volume of media taken out from the bottles before inoculation. Cultures were diluted every 3 or 4 d under the light and temperature treatments and every 2 d under the pH and CO₂ treatments, and maintained in exponential growth for 16 d at 28 µmol photons m⁻² s⁻¹, 11.0°C, pH 6.70, and 2.70 Pa CO₂, and for 12 d at 45 µmol photons m⁻² s⁻¹, 14.5°C, pH 7.40, and 6.40 Pa CO₂ with a minimum of 16 generations. The dilution volumes were about 4340 mL under the control, 28 µmol photons m⁻² s⁻¹ and 11.0°C treatments, 4390 mL under the 45 µmol photons m⁻² s⁻¹ and 14.5°C treatments, 4140 mL under the pH 7.40 and 6.40 Pa CO₂ treatments, and 3630 mL under the pH 6.70 and 2.70 Pa CO₂ treatments. Initial cell density was about 5000 cells mL⁻¹ at all treatments and final cell densities were 65,000–100,000 cells mL⁻¹ under the light and temperature treatments and 23,000–62,000 cells mL⁻¹ under the pH and CO₂ treatments (Fig. S1). Cultures were harvested 2 d after dilution in the control, the pH and CO₂ treatments, 3 d after dilution in the 45 µmol photons m⁻² s⁻¹ light treatment and 14.5°C temperature treatment, and 4 d after dilution in the 28 µmol photons m⁻² s⁻¹ irradiance and 11.0°C temperature treatments. Culture bottles were mixed three times per day at 09:00 h, 13:00 h, and 18:00 h. In the last day of the incubation at each light, temperature, pH, and CO₂ conditions, subsamples were taken for measurements of TA or DIC, pH_{Total}, cell concentrations, cellular contents of total particulate carbon (TPC), POC, nitrogen (PON) and phosphorus (POP), protein, RNA, carbohydrate, and pigment.

Carbonate chemistry measurements

Samples for determinations of pH_{Total} (total scale), TA, and DIC were syringe-filtered (0.22 µm pore size, Millex-GV filter), and the bottles were filled from bottom to top with overflow and closed immediately without a headspace. The pH_{Total} was measured immediately under the incubation temperatures using a pH meter which was calibrated with buffers (Tris•HCl, Hanna) at pH 4.01, 7.00 and 10.00. TA and DIC samples (250 mL) were treated with 100 µL saturated HgCl₂ solution

and stored in the dark at 4.0°C. TA and DIC samples were analyzed by potentiometric and coulometric titration, respectively, using a VINDTA 3C (Marianda) according to Dickson et al. (2007). Carbonate chemistry parameters were calculated using the CO2SYS program of Pierrot et al. (2006) with equilibrium constants of Roy et al. (1993). Initial CO₂ levels ranged from 40.80 to 42.42 Pa under the light, temperature, and pH treatments, and were 6.40 and 2.70 Pa under the CO₂ treatments (Table 1). Initial pH values were 8.03 or 8.04 under the light, temperature, and CO₂ treatments, and were 7.40 and 6.70 under the pH treatments. During the experiment, DIC concentrations decreased due to organismal activity by, on average, 0.57% to 38.91% under the various light, temperature, pH, and CO₂ treatments (Table 1). TA concentrations decreased by 2.30% to 7.27% under the light and temperature treatments, respectively. Correspondingly, CO₂ levels in the culture decreased by, on average, 13.40% to 40.73% under the various light, temperature, pH and CO₂ treatments. pH_{Total} values increased by less than 0.1 unit under all treatments (Table 1). The small changes in pH value under the pH and CO₂ treatments are due to HEPES buffering.

Cell density measurements

Twenty-five milliliter samples were taken daily at 13:00 h and fresh media with the same carbonate chemistry as in the initial treatment conditions were added as top-up. Cell density was measured using a Z2 Coulter Particle Count and Size Analyzer (Beckman Coulter). A flow cytometry (BD Accuri C6, MI) was used to measure cell density in the pH 6.70 and 2.70 Pa CO₂ treatments because coccospere volumes became too small to reliably measure in the Z2 Particle Counter. Variation in measured cell concentrations was ± 2% between the two methods. Growth rate (μ) was calculated for each replicate according to the equation: $\mu = (\ln N_t - \ln N_0)/t$, where N_t and N_0 refer to the cell concentrations in the last day and beginning of the experiment, respectively, and t was the growth period in days.

Elemental and macromolecular harvesting

After mixing, samples for determinations of TPC (250 mL), POC and PON (300 mL), POP (250 mL), carbohydrate (800 mL), chlorophyll *a* (Chl *a*), and carotenoids (100 mL) were obtained by filtering onto precombusted GF/F filters (at 450.0°C for 6 h, Whatman) 7 h after the onset of the light period (at 14:00 h). Samples for determination of protein (600 mL) and RNA (800 mL) were filtered onto PC filters (25 mm diameter, 0.6 µm pore size, Nuclepore, Whatman) also at 14:00 h. Elemental and macromolecular composition was sampled only at the end of the incubations. Under the pH 6.70 and 2.70 Pa CO₂ treatments, sampling volumes harvested for pigment, elemental, and macromolecular analyses were twofold larger than those under the other treatments.

Elemental analyses

To remove HEPES from the filters under the pH and CO₂ treatments, POC samples were rinsed three times with 30 mL of ASW (no nutrients added) with the same carbonate chemistry as in the treatment and without HEPES buffer. For POC and PON, samples were fumed with HCl for 12 h to remove inorganic carbon. TPC, POC, and PON samples were dried at 60.0°C for 12 h and analyzed using a Costech CHN analyzer. PIC quota was calculated as the difference between TPC and POC quota (Fabry and Balch 2010). To remove dissolved inorganic phosphorus from the filters, POP samples were rinsed three times with 5 mL 0.17 mol L⁻¹ Na₂SO₄. After that, 2 mL 0.017 mol L⁻¹ MgSO₄ solution was added onto filters and then POP samples were dried at 90.0°C for 12 h and combusted at 500°C for 6 h to remove POC, then cooled, and extracted by hydrolysis with 0.2 mol L⁻¹ HCl (Solórzano and Sharp 1980). Phosphorus concentrations were determined using the ammonium molybdate method (Chen et al. 1956) in a microplate reader (Thermo Scientific Varioskan Lux) using adenosine-5-triphosphate disodium trihydrate (ATP007, Bioshop) as a standard.

Macromolecular analyses

Carbohydrate samples were pretreated with 12.00 mol L⁻¹ of sulfuric acid (H₂SO₄) in the dark for 1 h, and then diluted by Milli-Q water for a final H₂SO₄ concentration of 1.20 mol L⁻¹. After a 5 min sonication and 30 s vortexing, samples were placed in a boiling water bath (90.0°C) for 3 h (Pakulski and Benner 1992). The concentration of monosaccharide was determined at 490 nm by phenol-sulfuric reaction with D-glucose as standard (Masuko et al. 2005).

Protein and RNA samples were immediately placed into 2 mL MP Biomedical tubes (Lysing Matrix D) containing large ceramic beads after harvesting and frozen by liquid nitrogen. Samples were stored in the dark at -80.0°C. Freeze-dried protein samples were extracted by a mixture of 0.5 mL 1X protein extraction buffer composed of lithium dodecyl sulfate (73.44 mmol L⁻¹), ethylene diamine tetraacetic acid (125.00 mmol L⁻¹), Tris base (28.07 mmol L⁻¹), Tris-HCl (21.10 mmol L⁻¹), glycerol (1085.89 mmol L⁻¹), and 4-(2-aminoethyl) benzenesulfonyl fluoride hydrochloride (0.10 mmol L⁻¹). Cells were lysed 4X1 min using a FastPrep system at 6.5 m s⁻¹. Between two cycles, samples were chilled in an ice bath for 2 min to prevent protein degradation. The samples were then centrifuged at 10,000 × *g* for 5 min (Spectrofluor 16M labnet). Extracted protein in the supernatant was quantified using the BCA assay with bovine gamma globulin as a standard, following Ni et al. (2016).

Total RNA was extracted by using TRIzol reagent (Invitrogen) and the RNeasy Plus Mini Kit (Qiagen). Samples were resuspended in 1.2 mL TRIzol reagent (Thermo Fisher Scientific) in lysing matrix D tubes (MP Biomedicals), homogenized by a FastPrep-24 machine (MP Biomedicals, 3 cycles, 8.0 m s⁻¹, 30 s, 3 min ice-chilling at interval), followed by

centrifugation at 12,000 × *g* for 3 min at 4°C (Eppendorf 5430R). RNA was extracted using a standard phenol-chloroform method (Chomczynski and Sacchi 1987). The aqueous phase was transferred to a gDNA eliminator column (from a RNeasy Plus Mini Kit) to remove the gDNA according to the manufacturer's protocol. RNA was purified by RNeasy Plus Mini Kit according to the manufacturer's protocol. Qiagen's RNase free DNase Set (an on column treatment) was further applied to remove contaminating genomic DNA according to the manufacturer's protocol. RNA was eluted into 40 µL RNase-free water, and its quality and content were analyzed using a microplate reader (Thermo Scientific Varioskan Lux). All RNA samples were tested for quality (A260/A280: 2.05–2.23; A260/A230: 2.10–2.57) (Sambrook et al. 1989).

Data analysis

ANOVA and Tukey post hoc tests were used to identify significant differences among three levels in each environmental factor (two levels for each of the light, temperature, pH, and CO₂ treatments, plus a common control for all treatments) for each of growth rate and the elemental and macromolecular ratios. Normality of residuals was checked using a Shapiro-Wilk test and a Levene test was conducted graphically to test for homogeneity of variances. A generalized least squares model was used to stabilize heterogeneity if variance was non-homogeneous. We described the relationship between measured parameters and growth rate as linear when the correlation coefficient (*r*) was larger than 0.80 in each environmental factor. The linear fitting and all statistical calculations were performed using R (R Core Team 2018) with the packages carData, lattice, and nlme.

Results

Elemental stoichiometry as a function of growth rate under different environmental conditions

Growth rates reduced to approximately $\frac{3}{4}\mu_{\max}$ and $\frac{1}{2}\mu_{\max}$ at the intermediate and low levels of light, temperature, pH, and CO₂, respectively, compared to the control (Fig. S1; Tables S2, S4). There is no universal pattern in elemental stoichiometry with growth rate (Fig. 1). PIC : POC is highest when growth rates are highest, and decreases with decreasing growth rates in each of the treatments (Fig. 1a). Except for the pH and CO₂ treatments, the shape and degree of change in PIC : POC with growth rate varies across the environmental treatments (Fig. 1a). C : N decreases linearly with decreasing growth rate in the light treatment ($r = 0.98$, $F = 195.50$, $p < 0.01$; Table S3), varies little with growth rate in the temperature treatment, and exhibits decreases in the pH and CO₂ treatments (Fig. 1b; Table S3). Both C : P and N : P increase with decreasing growth rate under the light treatment (both $r < -0.83$, $F > 22.92$, $p < 0.01$) but decrease similarly with decreasing growth rate in the temperature, pH and CO₂ treatments (all $r > 0.89$, $F > 38.02$, $p < 0.01$; Fig. 1c,d; Table S3).

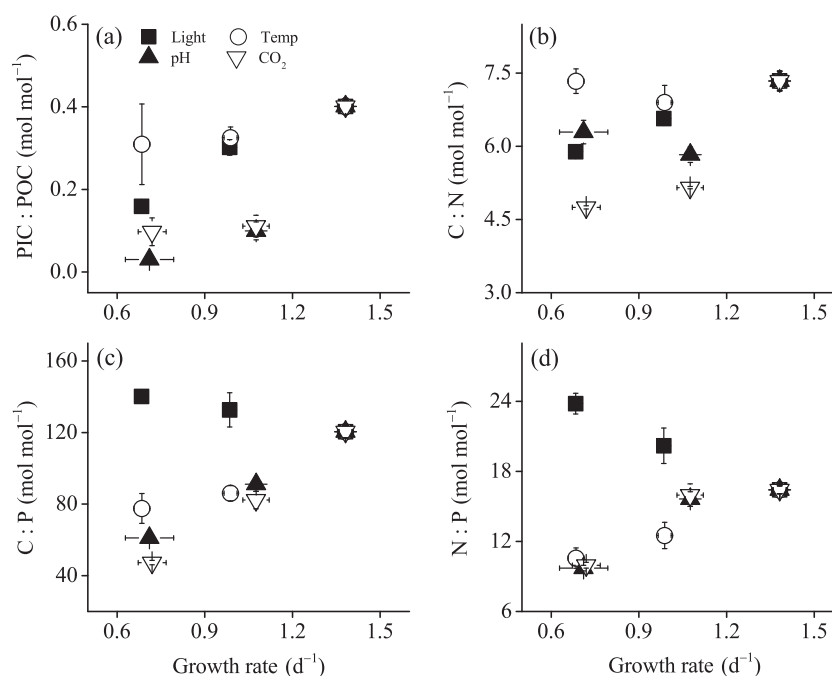


Fig. 1. The ratios of PIC : POC (a), C : N (b), C : P (c), and N : P (d) of *E. huxleyi* RCC1266 as a function of growth rate stimulated by a range of light, temperature (Temp), pH, and CO₂ conditions. Each point indicates mean ± SD of four replicates.

Macromolecular allocation as a function of growth rate under different environmental conditions

There are large differences in macromolecular allocation with growth rate across the light, temperature, pH, and CO₂ treatments (Fig. 2). In the light and CO₂ treatments, carbohydrate : POC decreases linearly with decreasing growth

rates (both $r > 0.92$, $F > 54.17$, $p < 0.01$; Fig. 2a; Table S3). In contrast, under the temperature and pH treatments, carbohydrate : POC declines moderately at $\frac{3}{4}\mu_{\max}$ relative to $\frac{1}{2}\mu_{\max}$ or μ_{\max} . Protein : POC does not change significantly with growth rate under the light treatment (Tukey HSD, $F = 3.62$, $p > 0.05$; Fig. 2b; Table S2), but under the

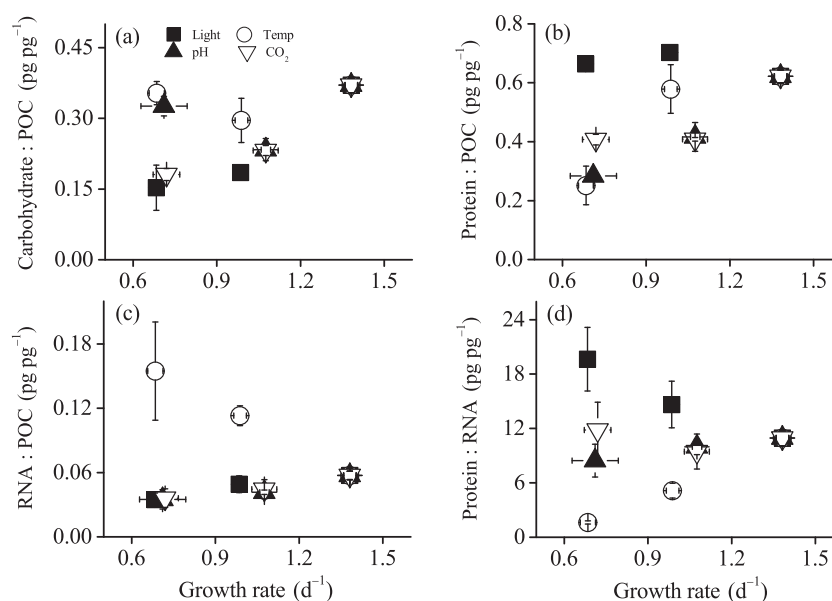


Fig. 2. Particulate organic carbon (POC) normalized carbohydrate (a), protein (b), and RNA (c), and protein : RNA (d) of *E. huxleyi* RCC1266 as a function of growth rate under the light, temperature (Temp), pH, and CO₂ treatments. Each point indicates mean ± SD of four replicates.

temperature treatment it decreases significantly between $\frac{3}{4}\mu_{\max}$ and $\frac{1}{2}\mu_{\max}$, while it is insignificantly different between μ_{\max} and $\frac{3}{4}\mu_{\max}$ (Table S2). Protein : POC appears to decrease linearly with decreasing growth rate in the temperature, pH, and CO₂ treatments ($r = 0.88$, $F = 23.68$, $p < 0.01$), although protein : POC is not significantly different between the $\frac{3}{4}\mu_{\max}$ and $\frac{1}{2}\mu_{\max}$ under the CO₂ treatment (Fig. 2b; Fig. S7h; Table S2). RNA : POC decreases linearly with decreasing growth rate in a very similar manner under the light, pH and CO₂ treatments (all $r > 0.83$, $F > 22.63$, $p < 0.01$; Fig. 2c; Table S3). In contrast, RNA : POC increases linearly with decreasing growth rates in the temperature treatment ($r = -0.84$, $F = 24.71$, $p < 0.01$; Fig. 2c; Table S3). Protein : RNA increases with decreasing growth rate in the light treatment, decreases with decreasing growth rates in the temperature treatment, and shows no significant change with growth rate under the pH or CO₂ treatments (Fig. 2d; Table S2).

Discussion

The GRH provides a mechanistic basis for modeling changes in macromolecular content and elemental composition (Elser et al. 2000, 2003; Inomura et al. 2020). Evidence supporting the GRH in phytoplankton is mixed. Here, we show that growth rate-associated RNA, protein, and carbon allocation strategies in *E. huxleyi* RCC1266 vary under different environmental conditions, but that physiological acclimation to specific conditions can, in part, explain deviations from the GRH and patterns in C : N : P. In this work, we compared RNA : POC and protein : POC allocation strategies and C : N : P under four different conditions (irradiance, temperature, pH, and CO₂) which were calibrated to reduce growth rate to about half and three-quarters of the maximum growth rate achieved in the control. RNA : POC and protein : POC increases with growth rate, for most but not all conditions (Fig. 2). Resource allocation to RNA and protein pools, and the resulting elemental ratios are linked to growth rate as predicted by the GRH, but the trends are modified by the effects of temperature on macromolecular synthesis rates and by the modulation of light, pH and CO₂ on protein requirements and carbon availability (Yvon-Durocher et al. 2015; Moreno and Martiny 2018). Below we discuss how physiological acclimation to temperature, irradiance, carbon limitation, and pH homeostasis can explain deviations in macromolecular and elemental composition relative to the expectations associated with the GRH (Figs. 1, 2).

Protein synthesis efficiency can be temperature sensitive, therefore the inverse relationship between cellular RNA content and RNA : POC with growth rate under the temperature treatment (Fig. 2c; Fig. S6j) likely indicates that *E. huxleyi* RCC1266 compensates for low temperature-induced decreases in protein synthesis efficiency by increasing ribosome content (Toseland et al. 2013; Schaum et al. 2018). Increases in RNA or

ribosome content under lower temperatures have been reported in the diatoms *Fragilariopsis cylindrus* and *Thalassiosira pseudonana* and there is some indication that this may occur in phytoplankton populations in the field (Toseland et al. 2013). The high RNA : POC under low growth rates in the low temperature treatment does not completely translate into low C : P or N : P (Figs. 1c,d, 2c). In *E. huxleyi* the relationship between RNA, cellular phosphorus or C : P with growth rate is complicated by condition-specific changes in intracellular phosphorus pools (Dyhrman 2016; Fig. 1c; Figs. S3m–p, S6i–l). RNA content typically contributes less than 40% of total organic phosphorus in *E. huxleyi* (Fig. S8m–p), and therefore other large pools such as polyphosphates can obscure the influence of RNA on C : P and N : P in *E. huxleyi* and other phytoplankton (Dyhrman et al. 2006; Nishikawa et al. 2006; Orchard et al. 2010).

Higher growth rates might be expected to require a higher protein : POC (Sterner and Elser 2002). Here, we find that protein : POC increases with temperature-, pH-, and CO₂-controlled increases in growth rate, but shows little change with light-induced increases in growth rate (Fig. 2b). The relatively elevated protein : POC under the two lower irradiance (28 and 45 $\mu\text{mol photons m}^{-1} \text{s}^{-1}$) treatments likely reflects an increased investment in pigment-protein complexes to increase light harvesting, and increased respiratory costs associated with their synthesis (cellular Chl *a* increases > twofold under these low irradiances (Fig. S2a)) (McKew et al. 2013). In *E. huxleyi*, and many other phytoplankton, higher light intensities are associated with lower pigment content, reducing the amount of light-induced photosystem II damage (Rokitta and Rost 2012; Li et al. 2021), and enhanced organic carbon storage in the form of carbohydrate (Fig. 2a) and lipid (Fernández et al. 1996). As a consequence, acclimation to higher irradiance is associated with elevated C : N and depressed N : P and protein : POC (Figs. 1b,d, 2b). The different patterns in protein : RNA as a function of growth rate across the environmental conditions examined are a consequence of distinct patterns in protein : POC and RNA : POC associated with acclimation to subsaturating irradiances and temperatures (Fig. 2). These results indicate that the relationship predicted by the GRH between protein : RNA and growth rate is modified by the physiological consequences of acclimation to environmental conditions.

Coccolithophores have an additional set of physiological changes induced by pH and CO₂ treatments due to calcification (Taylor et al. 2017) that can influence elemental and macromolecular composition. Both protein : POC and RNA : POC increase with growth rate across the pH and CO₂ treatments, resulting in no significant change in protein : RNA or N : P with growth rate in these treatments (Figs. 1d, 2d). In contrast, the pH and CO₂ treatments have different impacts on carbon allocation between POC and PIC (Figs. S3; Table S4). *E. huxleyi* acclimates to low CO₂ by lowering cellular Chl *a*, POC, and carbohydrate content (Table S4) and carbohydrate : POC

(Fig. 2a) and increasing carotenoid: Chl *a*. All these changes are indicative of inorganic carbon limitation of growth. Furthermore, *E. huxleyi* allocates more POC to protein synthesis and less to carbohydrate synthesis, leading to lower POC : PON and POC : POP (Fig. 1; Fig. S8d,h). POC synthesis is relatively more depressed under low CO₂ concentrations relative to the lowest pH conditions (Bach et al. 2013; Feng et al. 2018; Fig. S3g,h). Under the lowest pH treatment (highest H⁺ concentration), cellular POC and carbohydrate content increased > twofold, in comparison to the control (Figs. S3g, S6c). These results indicate that H⁺-driven inhibition in calcification may facilitate increased POC synthesis (Taylor et al. 2011). In other words, H⁺-driven increases in CO₂ uptake may compensate for DIC limitation for organic carbon fixation or carbohydrate synthesis (Rokitta et al. 2012; Kottmeier et al. 2016). Increases in POC content under pH 7.10 vs. pH 8.05 conditions has been reported in the diatom *Skeletonema costatum* (Thoresen et al. 1984).

The C : N : P of phytoplankton influences the biogeochemical cycling of these elements (Moreno and Martiny 2018). Traditionally ecosystem models use an invariant Redfield stoichiometry for modeling C : N : P. In recent years, modeling frameworks have been introduced to better understand and incorporate variable C : N : P (Daines et al. 2014). The next-generation ocean biogeochemical models require a better quantification of differences in C : N : P across major phytoplankton function types and across species within these groups under different environmental conditions (Daines et al. 2014; Yvon-Durocher et al. 2015; Inomura et al. 2020). Although the GRH does not strictly apply to *E. huxleyi*, there are clear patterns in how macromolecular and elemental compositions vary with growth rate under different environmental conditions that can be used to model the biogeochemical impacts of coccolithophores. Although much more work is required to examine a wider range of conditions and taxa, and examine interactions among changes in multiple conditions, we expect that incorporating acclimation strategies into phytoplankton growth models will improve our ability to model broad-scale patterns in ocean biogeochemistry.

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Conflict of Interest

The authors declared that they have no conflicts of interest.

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